

# Significant power factor improvement in a hybrid solid-liquid thermoelectric device formed by Sb:SnO<sub>2</sub> in contact with a chromium complex solution



*Castro-Ruiz. Sergio, Solís-De La Fuente. Mauricio, Márquez-García. Lourdes, Beltrán-Pitarch. Braulio, Mota-Babiloni. Adrián, Vidan. Francisco, García-Cañadas, Jorge*

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[Thermal and Electrical Systems Laboratory](http://teslab.uji.es) (<http://teslab.uji.es>)

Department of Industrial Systems Engineering and Design

Universitat Jaume I, Castelló de la Plana (Spain)

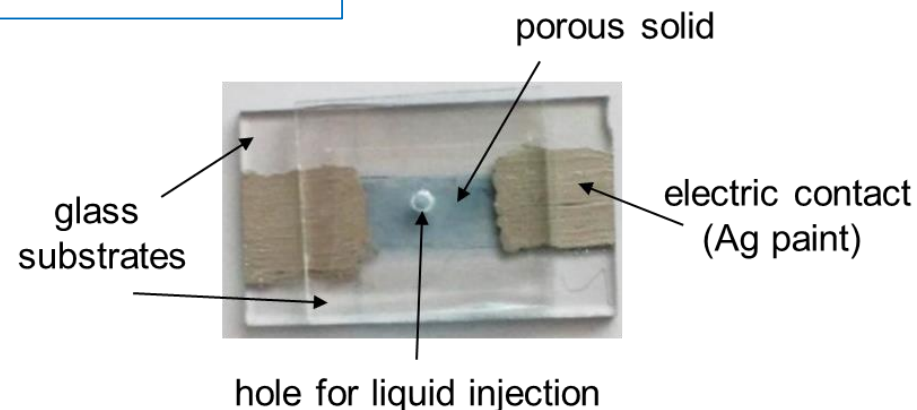
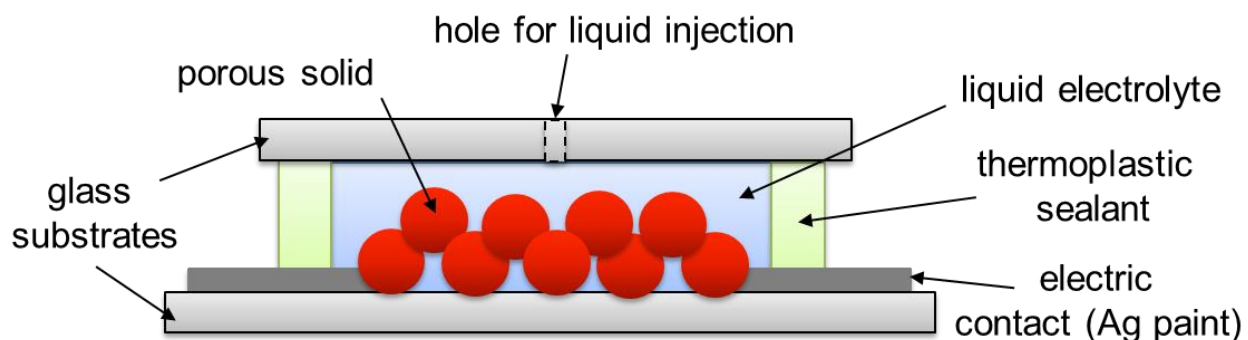
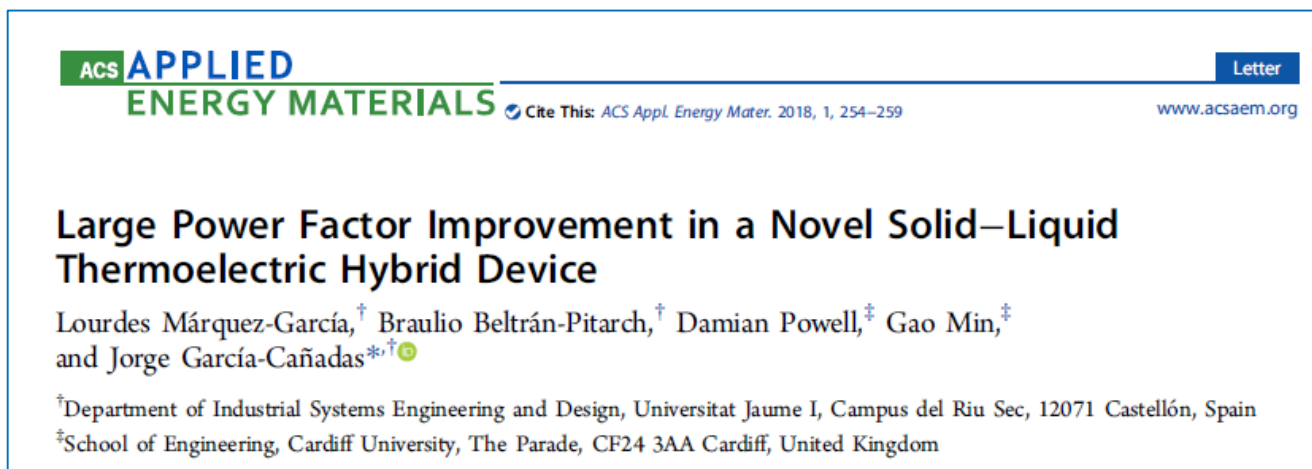
E-mail: [scastro@uji.es](mailto:scastro@uji.es)

## Outline

1. Introduction
2. Experimental part
3. Results
4. Summary
5. Acknowledgements



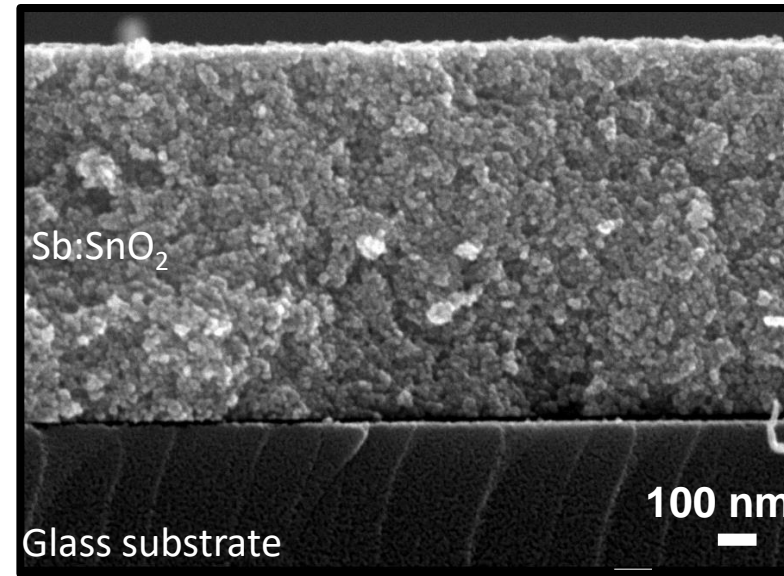
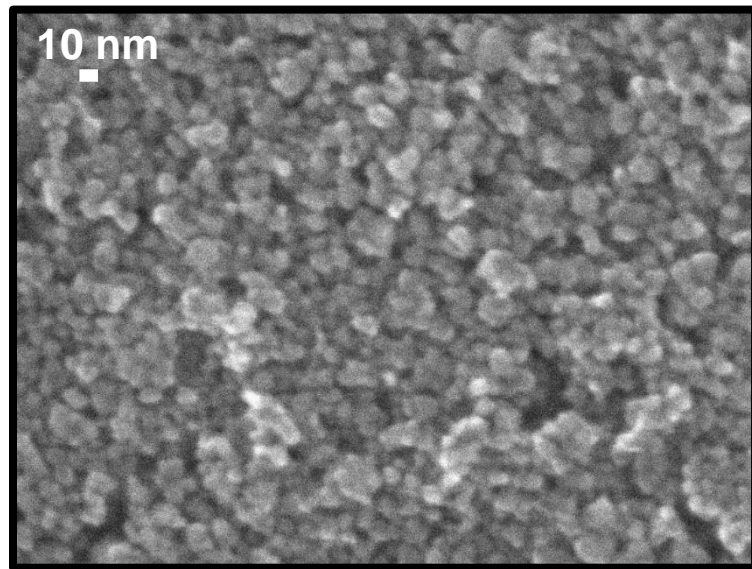
- ✓ A new hybrid solid-liquid thermoelectric (TE) device was published in 2018 by our group: TE solid which is permeated by a liquid electrolyte that was able to improve the power factor (PF) more than 3 times.



In this study, a **Cr complex solution** will be employed to prove if the approach can be extended to **this kind of electrolyte**.

### The porous solid: nanostructured $\text{Sb:SnO}_2$ (ATO)

- ✓ Prepared from commercial **colloidal water dispersion** (Keeling and Walker Ltd., UK). Deposited by **spin coating** (5 layers) onto a glass substrate and **annealed at 550 °C** for 45 min.
- ✓ Film is formed by interconnected **nanoparticles** of around **4 to 10 nm** diameter. The film thickness is  $\approx 1.0 \mu\text{m}$  (SEM) and it shows certain porosity ( $\approx 10\%$ ).

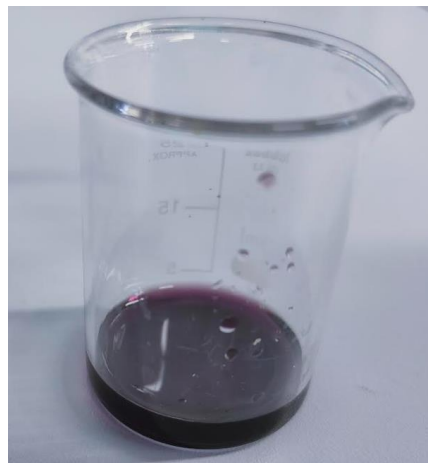
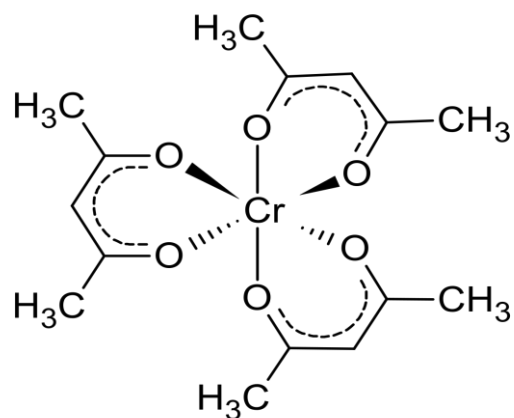


(SEM images)

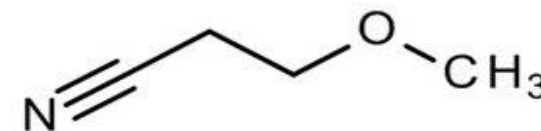
### The electrolyte: Chromium acetylacetonate ( $[Cr(acac)_3]$ ) in 3-MPN

- ✓ It is prepared by dissolving the **Cr complex** into the **3-methoxypropionitrile (3-MPN)** solvent to obtain a **0.1M solution**.

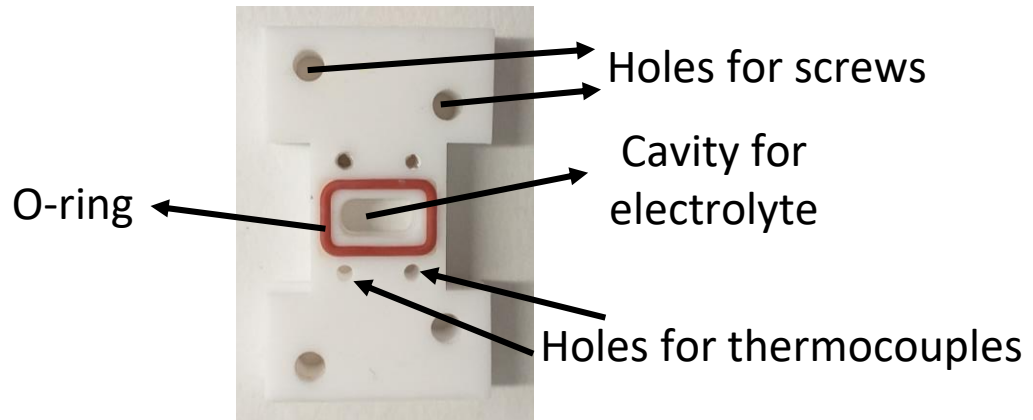
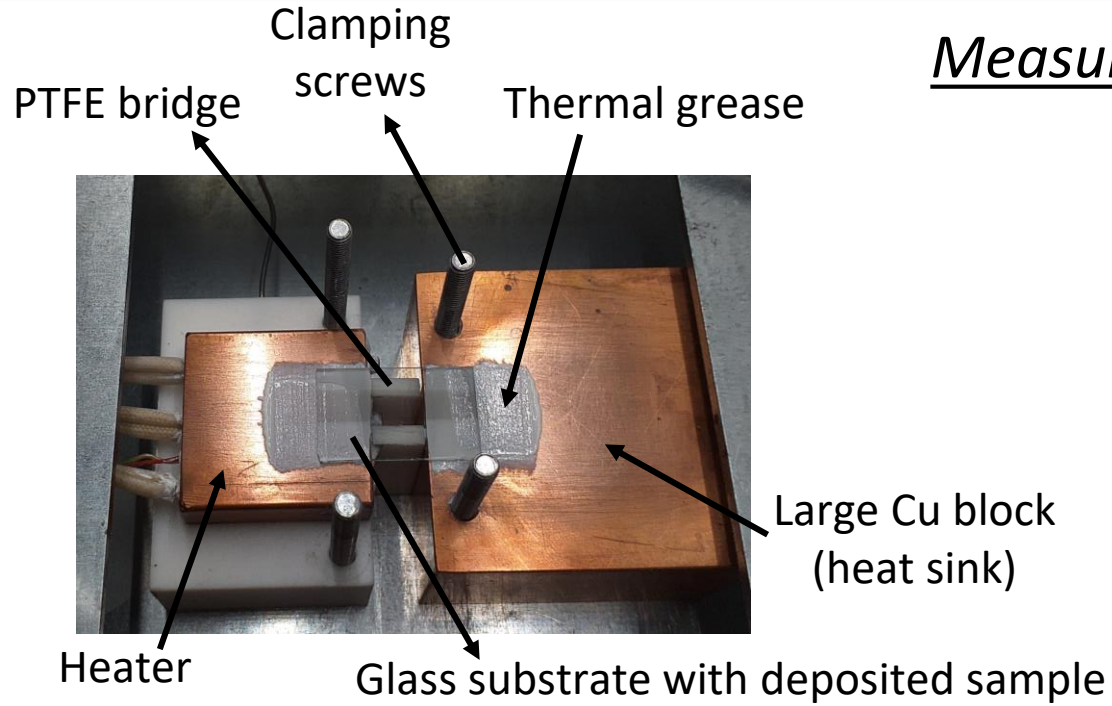
Chromium acetylacetonate  $[Cr(acac)_3]$



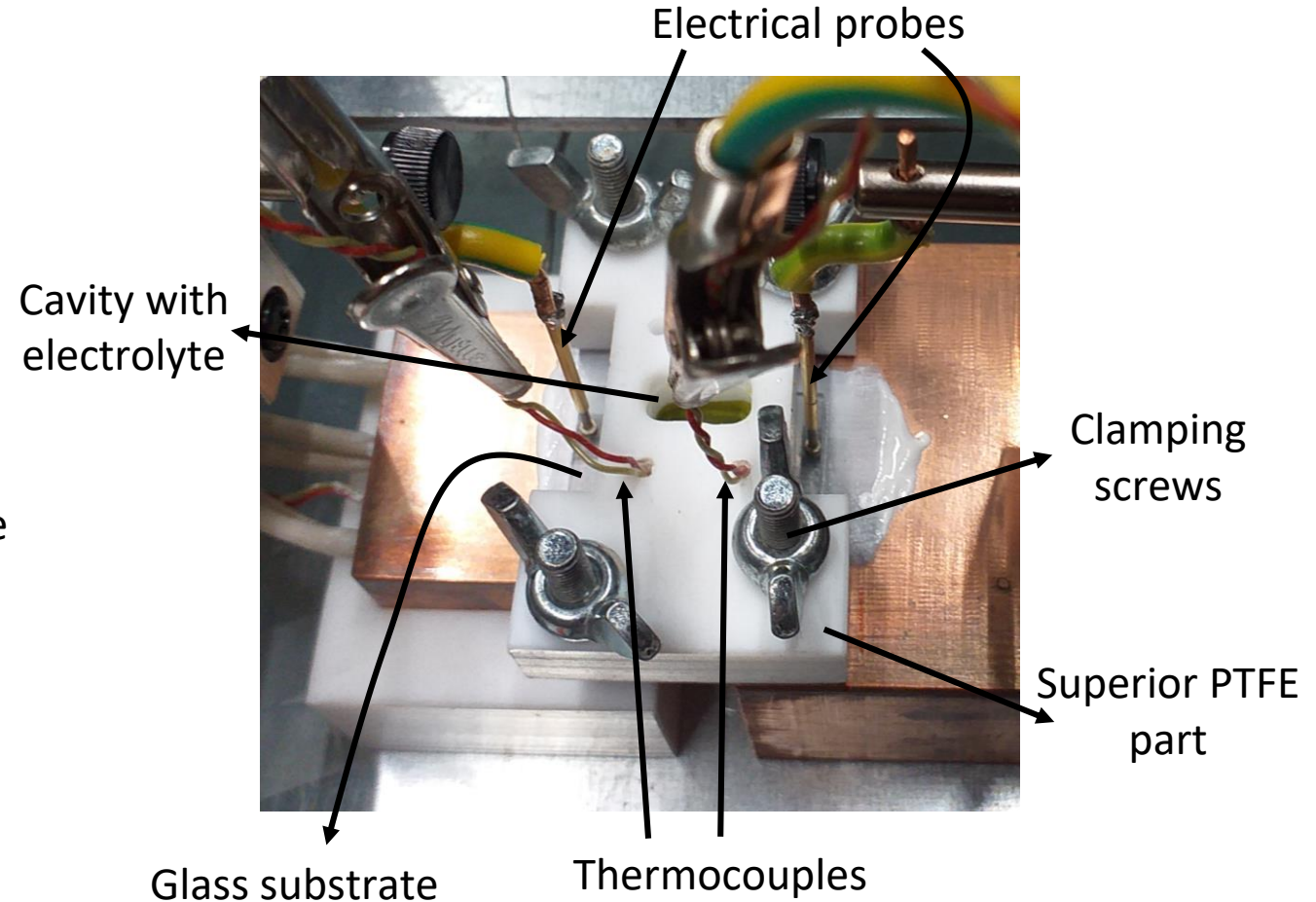
3-methoxypropionitrile



### Measurement setup

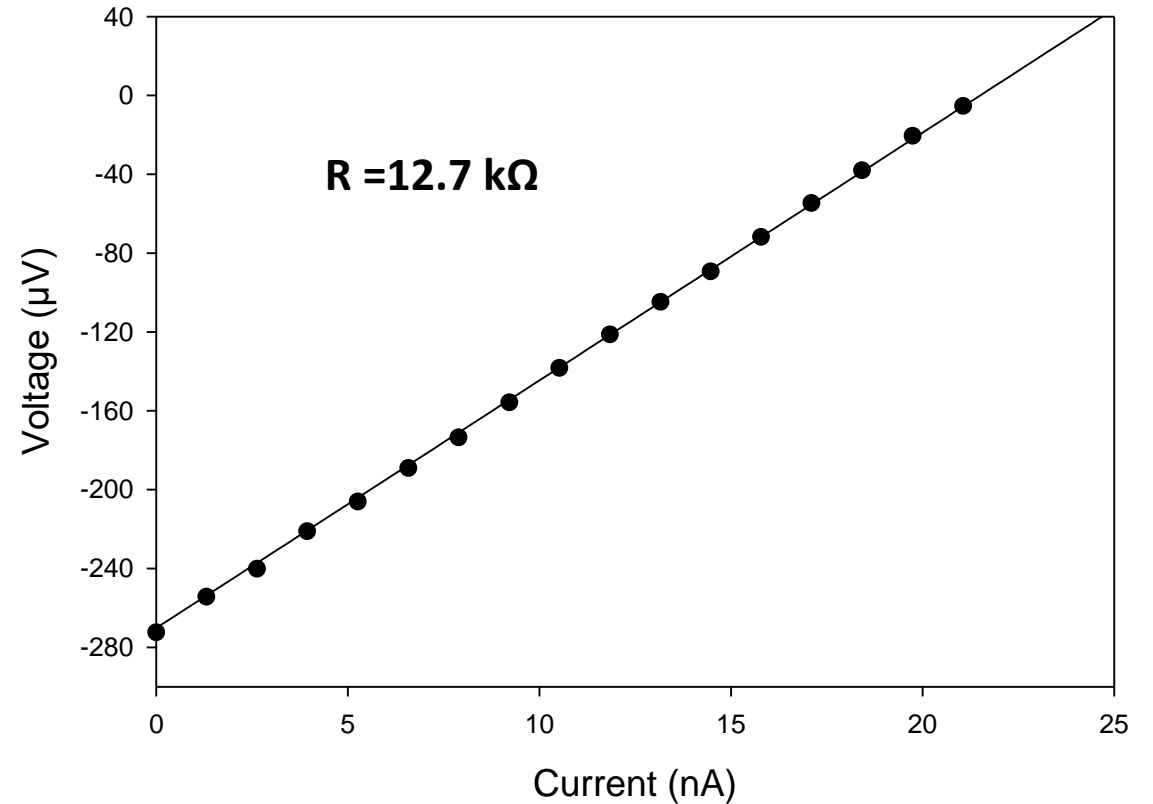
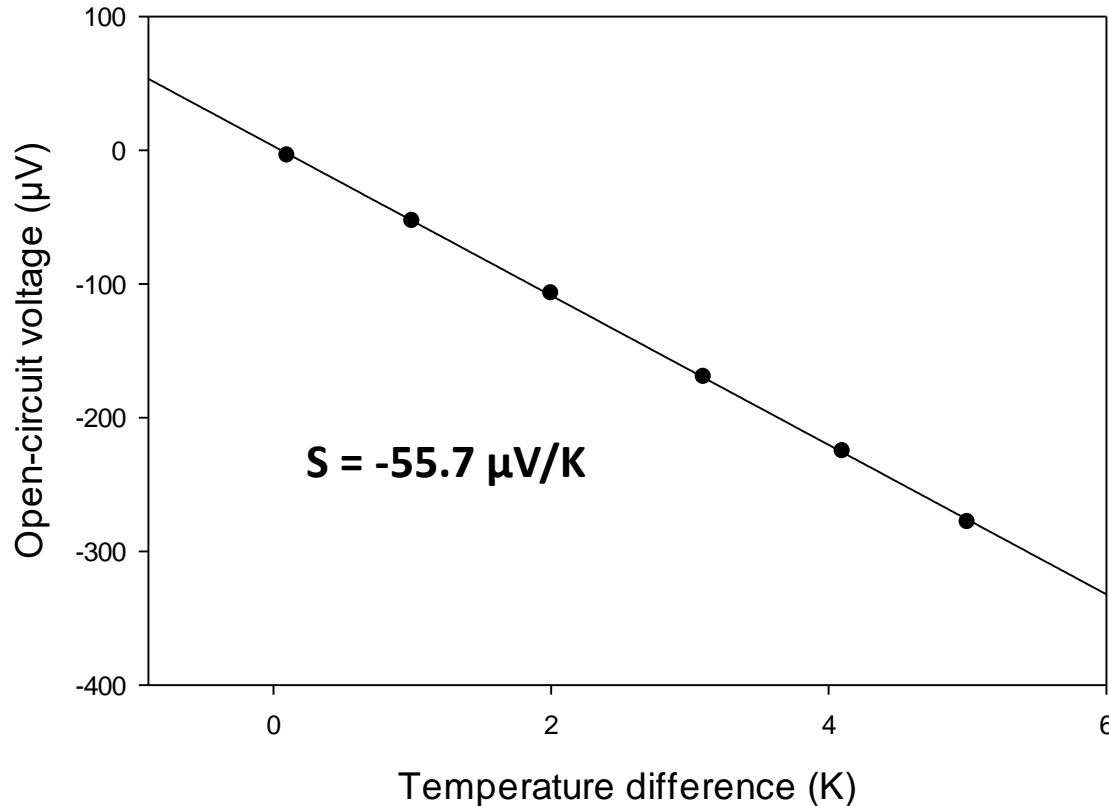


(bottom part of superior PTFE part)



### Thermoelectric measurements


- ✓ **Seebeck coefficient:** Extracted from the slope of the  $V_{oc} - \Delta T$  plot.
- ✓ **Device electrical resistance:** Extracted from the slope of the  $I - V$  curve at  $\Delta T = 5$  K.





(measurements of ATO film with no electrolyte)

- ✓ **ATO samples contacted by Ag paint** were tested **with and without** the presence of **0.1M Cr(acac)<sub>3</sub>** dissolved in **3-MPN** to evaluate the variation in the thermoelectric properties.

Sample	Seebeck coefficient ( $\mu\text{V/K}$ )			Device electric resistance ( $\text{k}\Omega$ )			$\text{PF}_{\text{with}}/\text{PF}_{\text{without}}$
	Without electrolyte	With electrolyte	S variation (%)	Without electrolyte	With electrolyte	R variation (%)	
S1	-55.7	<b>-40.1</b>	-27.9	12.7	<b>3.89</b>	-69.3	<b>1.71</b>
S2	-57.3	<b>-42.3</b>	-26.2	18.5	<b>7.56</b>	-58.9	<b>1.34</b>
S3	-59.4	<b>-51.2</b>	-13.9	10.2	<b>5.08</b>	-50.1	<b>1.48</b>

**Seebeck**  
 **22.7%**

**Resistance**  
 **59.4%**

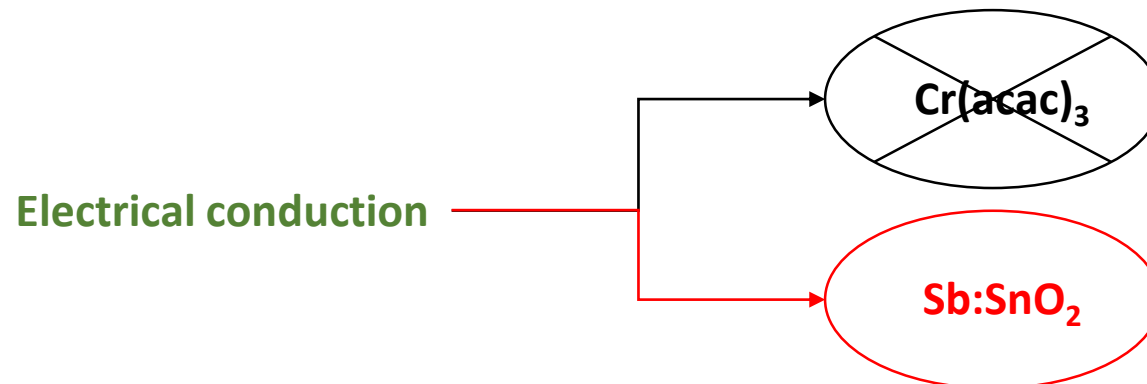
**Power Factor**  
 **1.51 times**



- ✓ **One sample** was fabricated **without ATO**, to evaluate if the **electrical conduction** occurs **through the electrolyte**.

Sample	Seebeck coefficient ( $\mu\text{V/K}$ )	Device electric resistance ( $\text{k}\Omega$ )
	With electrolyte	With electrolyte
Glass substrate with Ag paint contacts and $\text{Cr}(\text{acac})_3$	-	<b>4,580</b>

- ✓ Seebeck could not be measured due to the large resistance of the device. A value in the order of  $\text{M}\Omega$  was obtained for the device resistance, which proves that **electronic conduction is not possible through the electrolyte** and takes place mainly through the ATO.



- ✓ **Samples in contact with only 3-MPN** (no Cr complex) were tested to understand the system further.

Sample	Seebeck coefficient ( $\mu\text{V/K}$ )			Device electric resistance ( $\text{k}\Omega$ )			$\text{PF}_{\text{with}}/\text{PF}_{\text{without}}$
	Without electrolyte	With electrolyte	S variation (%)	Without electrolyte	With electrolyte	R variation (%)	
S4	-57.7	<b>-36.9</b>	-35.9	16.5	<b>9.40</b>	-43.1	<b>0.72</b>
S5	-57.1	<b>-43.1</b>	-24.7	8.70	<b>5.39</b>	-38.0	<b>1.00</b>
S6	-49.1	<b>-35.5</b>	-27.8	13.4	<b>8.88</b>	-33.4	<b>0.80</b>

**Seebeck**  
**29.5%**

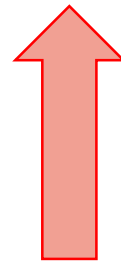
**Resistance**  
**38.2%**

**Power Factor**  
**0.84 times**

- ✓ **If  $\text{Cr}(\text{acac})_3$  is not present, PF improvement is not shown**, since the device resistance does not reach lower values [only 38% reduction, while 59% was obtained with  $\text{Cr}(\text{acac})_3$ ].

- ✓ **A hybrid system** formed by nanostructured **Sb:SnO<sub>2</sub>** permeated by a **Cr complex electrolyte** has been investigated to **improve** the thermoelectric **power factor**.
- ✓ More than **1.50 times improvement** in the power factor has been achieved due to a **59.4 % reduction of the device electric resistance** and a **22.7% drop** in the absolute value of the **Seebeck coefficient**.
- ✓ It was demonstrated that the **drop in the resistance is not due to the conduction through the electrolyte**.
- ✓ It has been shown that when **the Cr complex is not present in the system** (only **3-MPN as a electrolyte**), no PF improvement (0.82 times) is obtained.

$S^2\sigma$



*This electrolyte will be tested using state-of-the-art TE materials (e.g. Bi<sub>2</sub>Te<sub>3</sub>) within the EU UncorrelaTEd project*

- ✓ This project has received funding from the European Union's Horizon 2020 research and innovation programme under **grant agreement No 863222**.



<http://teslab.uji.es>

*E-mail: [scaastro@uji.es](mailto:scaastro@uji.es)*



<http://uncorrelated.uji.es/>

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